

Research on low temperature battery formulation technology

How does low temperature affect the performance and safety of lithium ion batteries?

Especially at low temperature, the increased viscosity of the electrolyte, reduced solubility of lithium salts, crystallization or solidification of the electrolyte, increased resistance to charge transfer due to interfacial by-products, and short-circuiting due to the growth of anode lithium dendrites all affect the performance and safety of LIBs.

Can high-throughput experiments be used in the research of low-temperature batteries?

Although many efforts have been made in the research of low-temperature batteries, some studies are scattered and cannot provide systematic solutions. In the future study, high-throughput experiments can be used to screen materials and electrolytes suitable for low-temperature batteries.

What factors limit the electrochemical performance of batteries at low temperatures?

At low temperatures, the critical factor that limits the electrochemical performances of batteries has been considered to be the sluggish kinetics of Li^+ .^{23,25,26} Consequently, before seeking effective strategies to improve the low-temperature performances, it is necessary to understand the kinetic processes in ASSBs.

What is a systematic review of low-temperature lithium-ion batteries?

In general, a systematic review of low-temperature LIBs is conducted in order to provide references for future research. 1. Introduction Lithium-ion batteries (LIBs) have been the workhorse of power supplies for consumer products with the advantages of high energy density, high power density and long service life .

Why do batteries need a low temperature?

However, faced with diverse scenarios and harsh working conditions (e.g., low temperature), the successful operation of batteries suffers great challenges. At low temperature, the increased viscosity of electrolyte leads to the poor wetting of batteries and sluggish transportation of Li^+ in bulk electrolyte.

How do electrolyte formulations affect low-temperature operation of LMBS?

As one of the fundamental components of a battery, the electrolyte greatly affects the low-temperature operation of LMBs. The electrolyte formulations have great influence on the ion conductivity, viscosity, interface features (e.g., compositions, structure and properties), solvation structure and de-solvation behaviors.

In this context, we discuss the microscopic kinetic processes, outline the challenges and requirements for low-temperature operation, highlight the materials and chemistry design strategies, and propose the future ...

In this article, we provide an overview of the low-temperature limiting mechanisms intrinsic to the lithium-ion battery chemistry, and then survey the field of next-generation battery chemistries ...

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In the last decade, the Li + solvation structure in electrolytes continues to gain attention, and an increasing number of research efforts are focused on understanding the impact of solvation structure on Li + transport and interphasial chemistry, aiming to improve battery performance by designing intermolecular interactions in the solvation structure [28, 29], which ...

Recently, great efforts have been devoted to the expeditions on low-temperature LMBs, which promote the establishment of fundamental recognitions of low-temperature battery chemistry, and proposal of various strategies to tame the low-temperature challenges.

Therefore, the rational formulation of electrolytes is significant for realizing superior low-temperature performance and broadening application niches of LIBs. Herein, we first discuss the...

Among various rechargeable batteries, the lithium-ion battery (LIB) stands out due to its high energy density, long cycling life, in addition to other outstanding properties. However, the capacity of LIB drops dramatically at low temperatures (LTs) below 0 °C, thus restricting its applications as a reliable power source for electric vehicles in cold climates and ...

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